

Aquabis[4-nitro-*N*-(quinolin-8-yl)benzene-sulfonamido- κ^2N,N']zinc(II)Luiz Everson da Silva,^a
Antonio Carlos Joussef,^a Sabine
Foro^b and Boris Schmidt^{b*}^aDepartamento de Química–UFSC, 88040-900
Florianópolis, SC, Brazil, and ^bClemens Schöpf-
Institut für Organische Chemie und Biochemie,
Technische Universität Darmstadt,
Petersenstrasse 22, D-64287 Darmstadt,
Germany

Correspondence e-mail: foro@tu-darmstadt.de

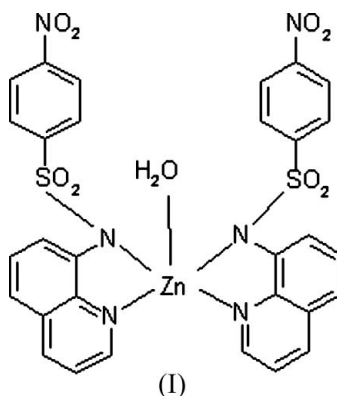
Key indicators

Single-crystal X-ray study
T = 299 K
Mean $\sigma(C-C)$ = 0.005 Å
R factor = 0.039
wR factor = 0.116
Data-to-parameter ratio = 12.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the title compound, $[Zn(C_{15}H_{10}N_3O_4S)_2(H_2O)]$, the asymmetric unit contains one half-molecule. The coordination of the Zn^{II} ion, which occupies a special position on a twofold axis, is distorted trigonal bipyramidal.

Received 6 June 2006

Accepted 3 July 2006

Comment

There is increasing evidence that interactions between the amyloid beta peptide and metal ions such as copper and zinc may assist in the transformation of this healthy soluble peptide to a neurotoxic fibrillary form responsible for Alzheimer's disease (Miller *et al.*, 2005; Cardoso *et al.*, 2005; Ii, 1995). As part of our efforts to search for metal chelators as potential probes for neuroprotection in neurodegenerative diseases (da Silva *et al.*, 2006a,b,c,d) the structure of the title compound, (I) (Fig. 1), has been determined.The asymmetric unit of (I) contains one half-molecule. The other half of the complex is related by a C_2 axis running through the Zn^{II} ion and the water O atom. The Zn^{II} ion has a distorted trigonal-bipyramidal geometry formed by two quinoline N and two sulfonamide N atoms and the O atom of the water molecule (Table 1). The angle between the least-squares planes of the quinoline unit and the benzene ring is $89.27(8)^\circ$. The water H atoms are shared between the water O atom and sulfonyl O atoms, linking the molecules in chains along the *c* axis (Fig. 2 and Table 2).

Experimental

The organic ligand was synthesized as reported previously (da Silva *et al.*, 2005) and the title compound was prepared according to the method of Macías *et al.* (2003). Single crystals of (I) suitable for X-ray data collection were obtained after two days from a methanol solution.

Crystal data

[Zn(C₁₅H₁₀N₃O₄S)₂(H₂O)]
M_r = 740.03
 Monoclinic, *C*2/*c*
a = 31.407 (6) Å
b = 10.120 (1) Å
c = 10.342 (1) Å
 β = 106.94 (2)°
V = 3144.5 (8) Å³

Z = 4
D_x = 1.563 Mg m⁻³
 Cu Kα radiation
 μ = 2.89 mm⁻¹
T = 299 (2) K
 Needle, yellow
 0.50 × 0.10 × 0.05 mm

Data collection

Nonius CAD-4 diffractometer
 ω/2θ scans
 Absorption correction: ψ scan
 (North *et al.*, 1968)
T_{min} = 0.326, *T_{max}* = 0.869
 3477 measured reflections
 2798 independent reflections

2319 reflections with *I* > 2σ(*I*)
R_{int} = 0.020
 θ_{max} = 66.9°
 3 standard reflections
 frequency: 120 min
 intensity decay: 1%

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.039
wR(*F*²) = 0.116
S = 1.05
 2798 reflections
 221 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

w = 1/[σ²(*F_o*²) + (0.0708*P*)²
 + 2.3429*P*]
 where *P* = (*F_o*² + 2*F_c*²)/3
 (Δ/σ)_{max} = 0.002
 Δρ_{max} = 0.30 e Å⁻³
 Δρ_{min} = -0.74 e Å⁻³

Table 1

Selected geometric parameters (Å, °).

N1—Zn1	2.187 (2)	Zn1—O1W	2.011 (3)
N2—Zn1	2.091 (2)		
O1W—Zn1—N2	124.48 (6)	N2 ⁱ —Zn1—N1	99.52 (8)
N2 ⁱ —Zn1—N2	111.04 (12)	N2—Zn1—N1	76.51 (8)
O1W—Zn1—N1	93.44 (6)	N1—Zn1—N1 ⁱ	173.12 (12)

Symmetry code: (i) -*x* + 1, *y*, -*z* + ²/₃.

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1W—H1W...O1 ⁱⁱ	0.87 (2)	2.44 (3)	2.935 (2)	117 (3)

Symmetry code: (ii) -*x* + 1, -*y*, -*z* + 2.

The symmetry-independent water H atom was located in a difference Fourier map and was refined with restrained geometry, *viz.* O—H restrained to 0.85 (2) Å and H...H restrained to 1.365 (2) Å, thus leading to an H—O—H angle of 109 (3)°. All other H atoms were placed in calculated positions and treated as riding, with C—H = 0.93 Å. All H atoms were refined with *U*_{iso}(H) = 1.2*U*_{eq}(parent atom).

Data collection: *CAD-4-PC* Software (Nonius, 1996); cell refinement: *CAD-4-PC* Software; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

The authors thank Professor Dr Hartmut Fues, FG Strukturforschung, FB Material- und Geowissenschaften, Technische Universität Darmstadt, for diffractometer time.

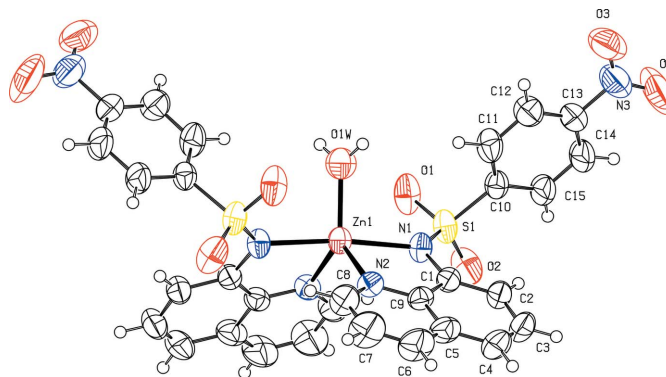


Figure 1

The molecular structure of the title compound, showing the atom labelling and displacement ellipsoids drawn at the 50% probability level. Unlabelled atoms are related to labelled atoms by the symmetry operation (-*x* + 1, *y*, -*z* + ²/₃).

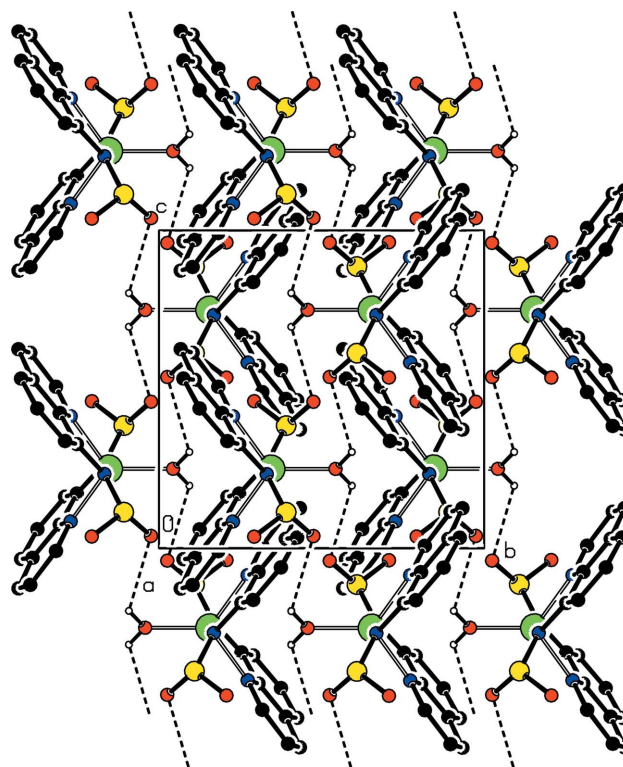


Figure 2

The molecular packing of the title compound, viewed along the *a*-axis direction, with hydrogen bonds shown as dashed lines. H atoms not involving in hydrogen bonding have been omitted.

References

- Cardoso, S. M., Rego, A. C., Pereira, C. & Oliveira, C. R. (2005). *Neurotoxicol. Res.* **7**, 273–281.
 Ii, K. (1995). *Drugs Aging*, **7**, 97–109.
 Macías, B., García, I., Villa, M. V., Borrás, J., Castiñeiras, A. & Sanz, F. (2003). *Z. Anorg. Allg. Chem.* **629**, 255–260.
 Miller, L. M., Wang, Q., Telivala, T. P., Smith, R. J., Lanzirrotti, A. & Miklossy, J. (2005). *J. Struct. Biol.* **152**, 204–211.
 Nonius (1996). *CAD-4-PC* Software. Version 1.2. Nonius, Delft, The Netherlands.
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.

- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Silva, L. E. da, Joussef, A. C., Foro, S. & Schmidt, B. (2005). *Acta Cryst.* **E61**, o3778–o3779.
- Silva, L. E. da, Joussef, A. C., Foro, S. & Schmidt, B. (2006a). *Acta Cryst.* **E62**, m516–m517.
- Silva, L. E. da, Joussef, A. C., Foro, S. & Schmidt, B. (2006b). *Acta Cryst.* **E62**, m518–m519.
- Silva, L. E. da, Joussef, A. C., Foro, S. & Schmidt, B. (2006c). *Acta Cryst.* **E62**, m912–m913.
- Silva, L. E. da, Joussef, A. C., Foro, S. & Schmidt, B. (2006d). *Acta Cryst.* **E62**, m999–m1001.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Stoe & Cie (1987). *REDU4*. Version 6.2c. Stoe & Cie GmbH, Darmstadt, Germany.